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### A New Liquid Crystalline Substance Exhibiting A Reextrant Nematic Phase and Two Low-Temperature Smectic Phases

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A NEW LIQUID CRYSTALLINE SUBSTANCE EXHIBITING  
A REENTRANT NEMATIC PHASE AND TWO LOW-TEMPERATURE  
SMECTIC PHASES

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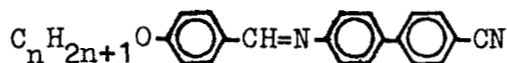
Abstract: A few members of the homologous  
4-n-alkyloxy-benzylidene-amino-4'-cyanobiphe-  
nyls were synthesized. The C<sub>7</sub>-compound exhi-  
bits a reentrant nematic phase and at lower  
temperatures smectic B and smectic E phases.

We have synthesized five members of the homolo-  
gous 4-n-alkyloxy-benzylidene-amino-4'-cyanobiphe-  
nyls in analogy to the 4-cyanobiphenyl 4'-n-alkyloxy-  
benzoates <sup>1</sup>. The substances were obtained by con-  
densation of 4-n-alkyloxybenzaldehydes with 4-ami-  
no-4'-cyanobiphenyl. The products were purified  
by recrystallization from ethanol or amylalcohol.

The transition temperatures determined with a  
polarizing microscope are given in table 1.

The C<sub>5</sub>-compound shows S<sub>E</sub>-S<sub>B</sub>-N polymorphism.

TABLE 1



n	cr	$S_E$	$S_B$	$N_{re}$	$S_A$	N	is.
5	.	119 . (76)	.	-	-	299	.
6	.	92 -	.	111 -	-	280	.
7	.	73 . 95	.	111 . 140	.	212 . 273	.
8	.	68 . 100	-	-	.	243 . 267	.
9	.	76 . 99	-	-	.	249 . 258	.

cr : solid crystal      The numbers are the  
 $S_A$ ,  $S_B$ ,  $S_E$ : smectic A, B or C      transition tempera-  
 N : nematic      tures ( $^{\circ}\text{C}$ ). Brackets  
 $N_{re}$  : reentrant nematic      denote monotropic  
 is. : isotropic      phases.

In the next homologue ( $C_6$ ) we could not detect a  $S_E$  phase, although the  $S_B$  phase was supercooled to room temperature. In the  $C_7$ -compound, below the temperature range of the reentrant nematic ( $N_{re}$ ) phase, two smectic low temperature phases -  $S_E$  and  $S_B$  - could be observed. In the  $C_8$  and  $C_9$  homologues, the  $S_B$  phase and the  $N_{re}$  phase disappear and a polymorphism  $S_E$ ,  $S_A$ , N occurs.

In table 2 the transition enthalpies of the  $C_7$  derivative measured with a differential scann-

ing calorimeter (DSC 2, Perkin-Elmer) are listed. It is remarkable that the enthalpy for the transition  $N_{re} \rightarrow S_A$  is very low compared with the enthalpy of the  $S_A \rightarrow N$  transition or with the corresponding values of other substances with reentrant nematic phases<sup>2,3</sup>.

TABLE 2 Transition Enthalpies ( $J\ mol^{-1}$ ) of 4-n-Heptyloxy-benzylidene-amino-4'-cyanobiphenyl

transition	transition temperature/ $^{\circ}C$	$\Delta H/J\ mol^{-1}$
cr I - cr II	65	23 500
cr II - $S_E$	73	4 500
$S_E$ - $S_B$	95	395
$S_B$ - $N_{re}$	111	3 400
$N_{re}$ - $S_A$	140	4 - 5
$S_A$ - N	212	37
N - is	273	1 000

On microscopic observation the  $S_B$  phase appears as a homeotropic texture, a fan-shaped texture or a typical mosaic texture depending on the experimental conditions. The  $S_E$  phase often occurs as a fan-shaped texture or a paramorphic mosaic texture.

The identification of the low temperature smectic phases was carried out by miscibility in-

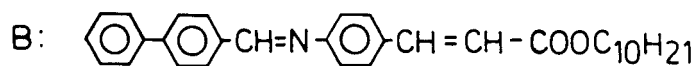
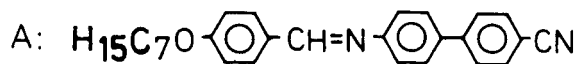
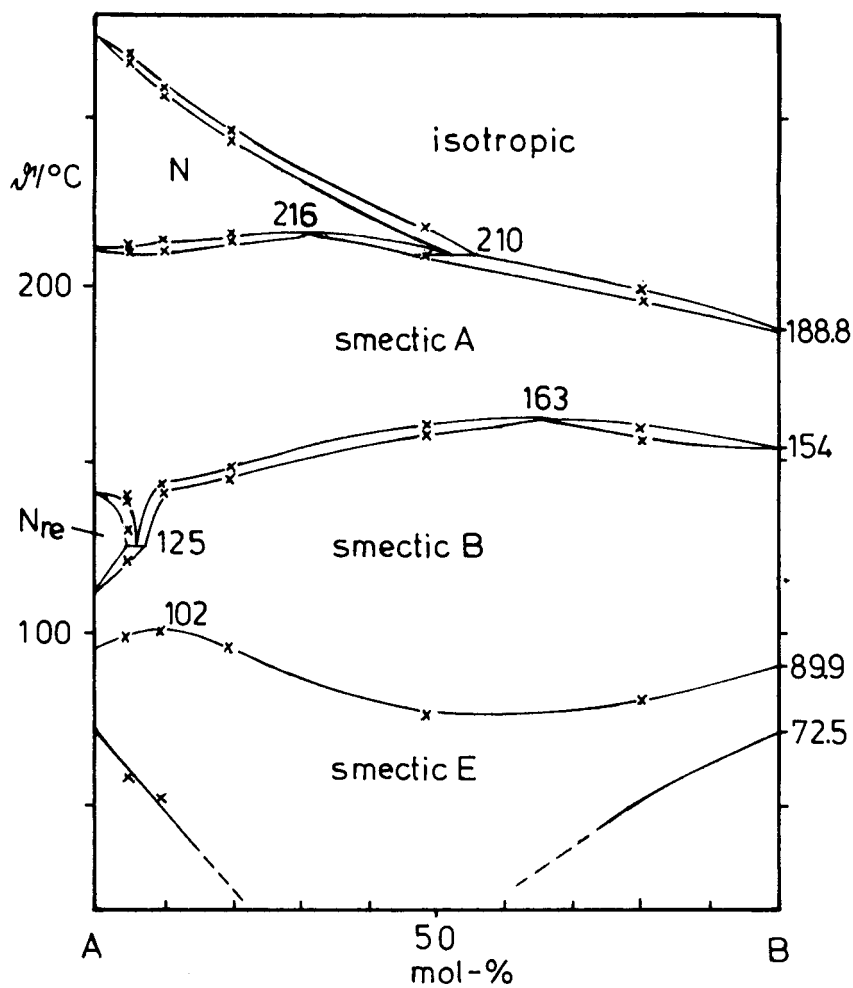


FIGURE 1

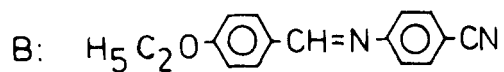
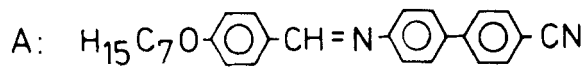
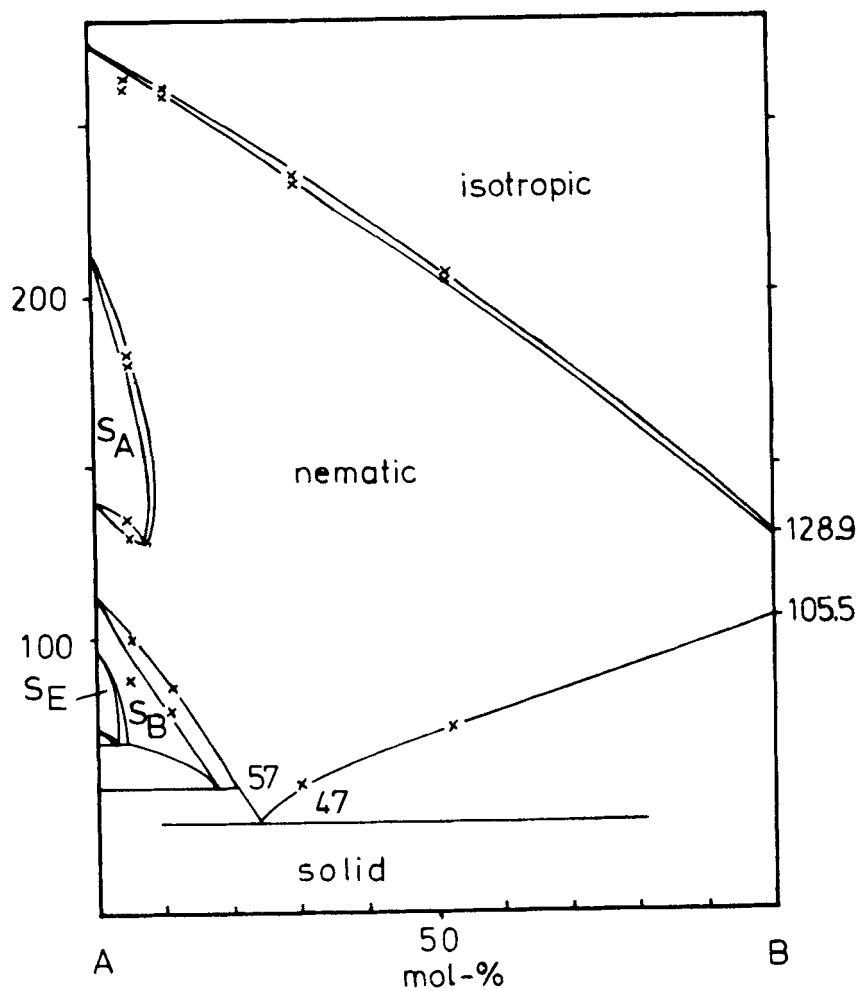


FIGURE 2

vestigations with suitable reference substances. The diagrams of state were determined by the contact method,<sup>4</sup> completed by the study of singular concentrations.

Figure 1 shows that the two smectic low temperature phases of 4-n-heptyloxy-benzylidene-amino-4'-cyanobiphenyl are completely miscible with the  $S_E$  and  $S_B$  phases of n-decyl 4-[4-phenyl-benzylidene-amino]-cinnamate<sup>5</sup>. The  $N_{re}$  phase exists only in a small concentration range. The eutectic temperature and also the melting points of the mixtures could not be determined because the samples did not crystallize.

As shown in figure 2 the nematic phase of ethoxybenzylidene-amino-4'-cyanoaniline<sup>6</sup> is completely miscible with the nematic high-temperature phase as well as with the reentrant nematic phase.

#### REFERENCES

1. F.Hardouin, A.M.Levelut, N.H.Tinh and G.Sigaud, Mol.Cryst.Liqu.Cryst.(Letters) 56, 35 (1979)
2. F.Hardouin, G.Sigaud, M.F.Achard and H.Gasparoux, Solid State Commun. 30, 365 (1979)
3. W.Weißflog, N.K.Sharma, G.Pelzl and D.Demus, Kristall und Technik, in preparation
4. H.Sackmann and D.Demus, Mol.Cryst.Liqu.Cryst. 21, 239 (1973)
5. A.Biering, D.Demus, G.W.Gray and H.Sackmann, Mol.Cryst.Liqu.Cryst. 28, 275 (1974)
6. E.Froelich, Dissertation Halle/S. 1910